

US EPA ARCHIVE DOCUMENT

New project:

**Impact of climate change on air
quality in the U.S.:
Global- and regional-scale models
for ozone and mercury**

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with results from previous EPA-STAR project:

Models and measurements for investigating atmospheric transport and photochemistry of mercury

Gerald J. Keeler and Sanford Sillman

(EPA R-82979901-0)



Critical Issues for modeling

- Predicting future air quality in response to changes in climate+emissions (in coordination with HTAP)
- What measurements can identify the impact of global changes in air quality?
- Impact of global background p'chem versus episodic transport
- Does ozone affect atmospheric mercury?
- Model improvements: integrated gas, aqueous and aerosol chemistry

Models

- Climate:
NCAR Finite-Volume Community Climate Model (FVCCM)
w/ IPCC emission scenarios for 2050
including aerosol-climate coupling
- Global photochemistry/transport:
IMPACT (LLNL/Michigan)
including hybrid dynamical representation of
nitrate/ammonia aerosols (Feng, 2006)
- Regional: modified CMAQ for eastern USA

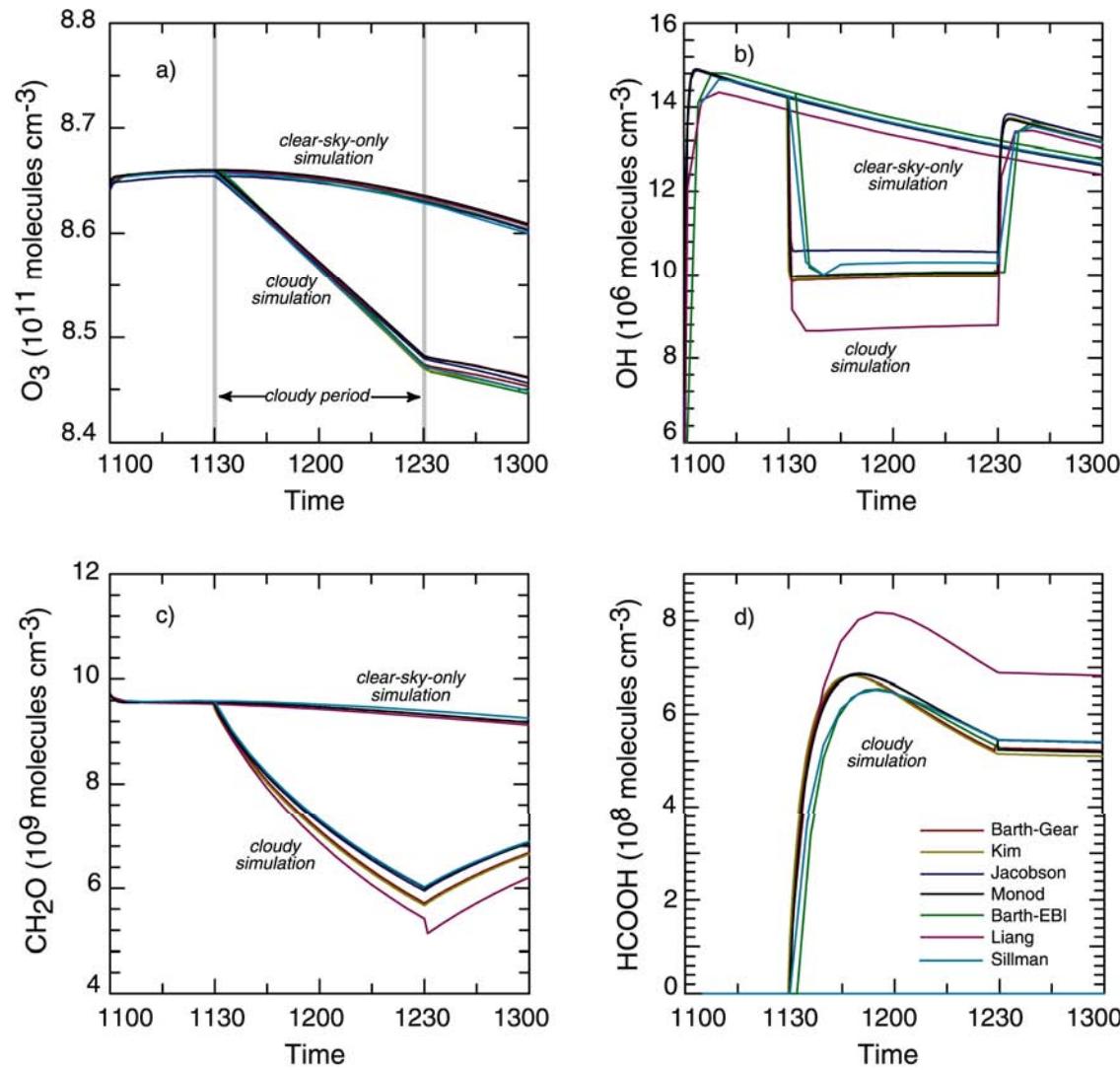
Modifications to CMAQ

- Integrated solver for combined gas+aqueous photochemistry (simultaneous solution)
- 70 aqueous reactions, 26 Hg reactions, 300+ gas-phase reactions (for O₃, NO_x, VOC, SO₄, halogens, etc)
- Aqueous solver tested in model intercomparison (Barth et al., 2003)
- Improved representation of j-values in clouds, replacing approximate format in original CMAQ
- Future: integrate gas+aqueous+aerosol chemistry
- Same gas+aqueous chemistry in global model

Numerical solution for gas+aqueous photochemistry

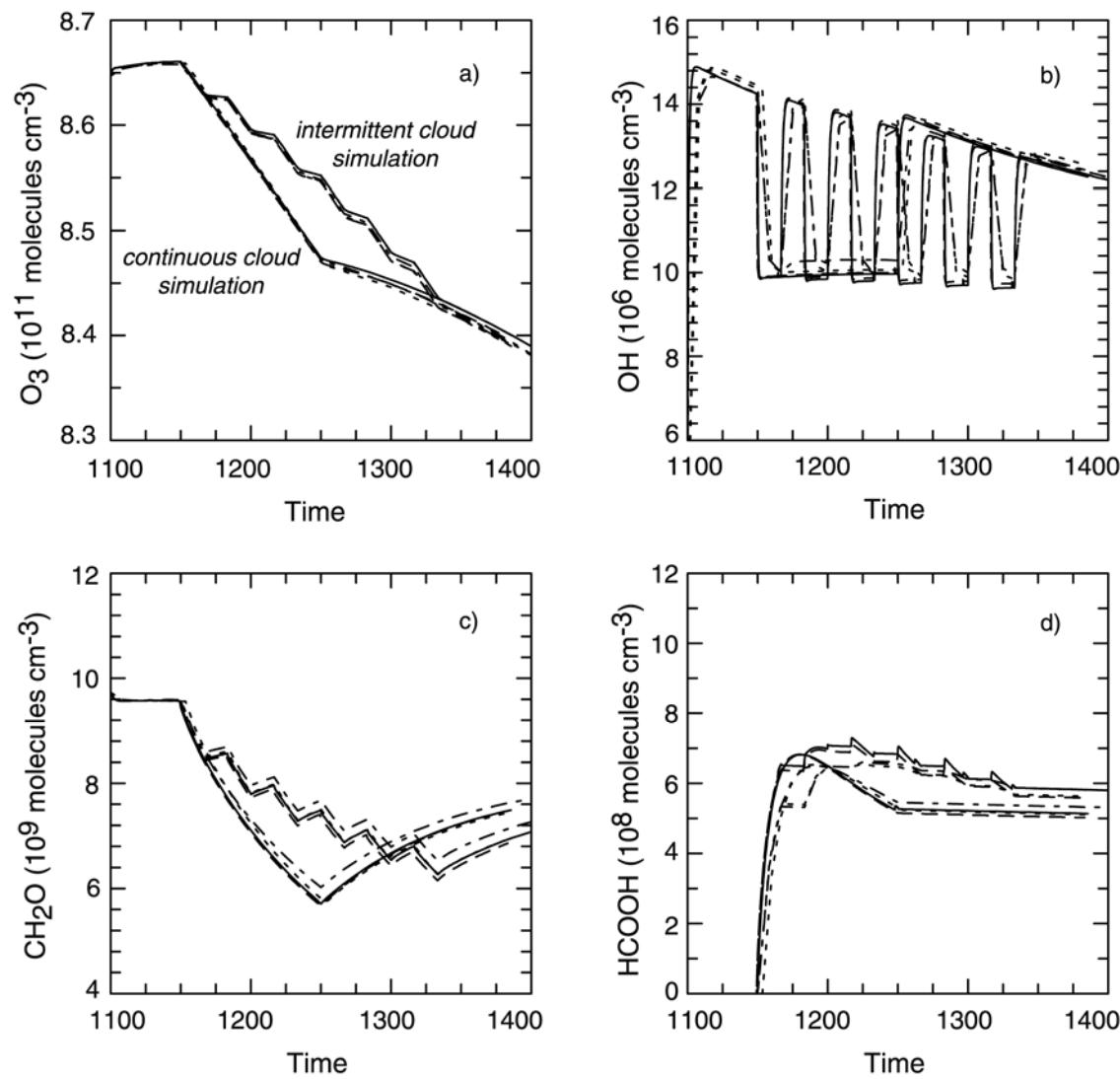
- Implicit (reverse Euler) solution with extension for exponential decay in remote locations
- Calculates gas-aqueous partitioning, gas-phase and aqueous p'chem.
- Fast solver for j-values in clouds
- Direct input of individual reactions
- Integrated into CMAQ
- Tested in model intercomparison (Barth 2003)
(including effect of intermittent clouds)

Model intercomparison (Barth et al., 2003)



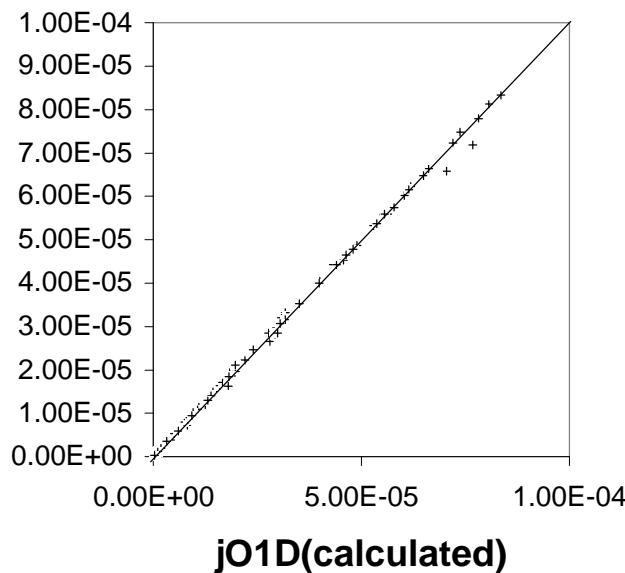
Intermittent clouds

(Barth et al., 2003)

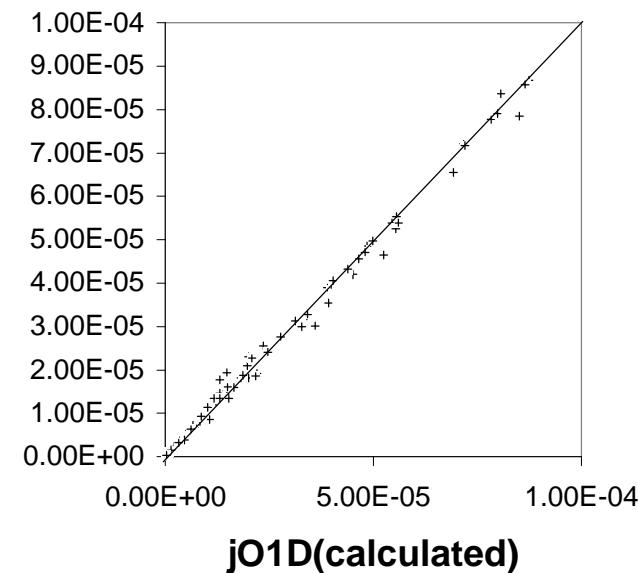


Evaluation of photolysis parameterization (vs. TUV model, Madronich and Flocke)

Single cloud layer



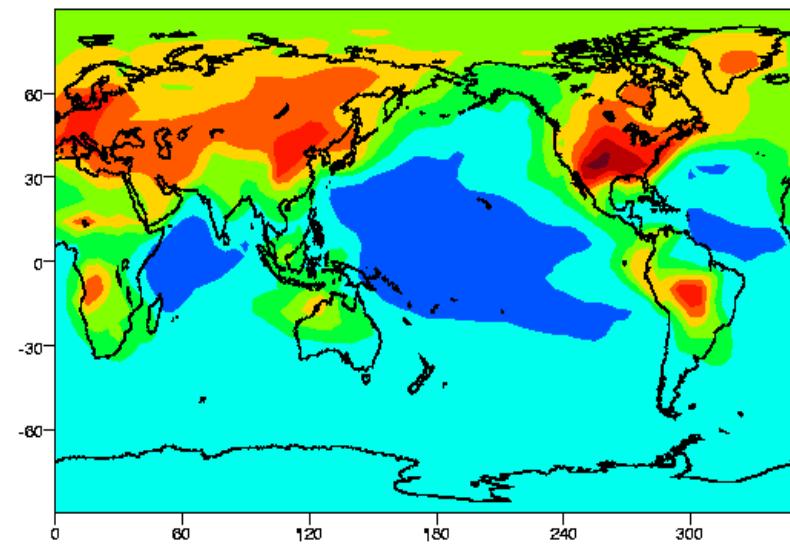
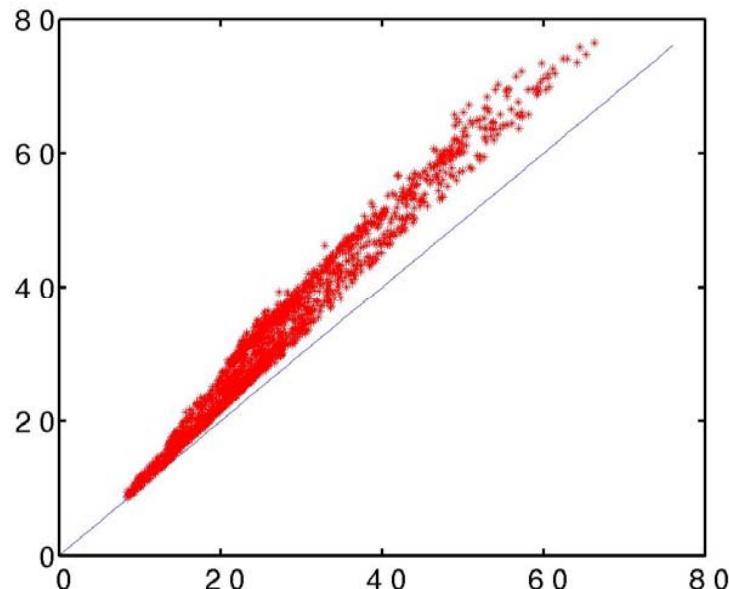
Two cloud layers



Previous result: global O₃

- Inclusion of aromatics, terpenes and isoprene nitrates affects model O₃:
20% increased O₃
30% increased PAN
(*Ito et al.*, 2006, consistent with *von Kuhlmann et al.*, 2005, etc.)
- Large changes in methyl glyoxal, hydroxyacetone (test vs. measurements from Spaulding et al., 2003) - possible implication for aerosol formation

O₃ (Extended - Base) Chemistry



Measurements vs models w/ base and extended chemistry

Species	Measured (Spaulding) (ppb)	Base model (ppb)	Extended model (ppb)
Glycol-aldehyde	0.63	0.06	0.30
Hydroxy-acetone	0.38	0.16	0.68
Methyl glyoxal	0.12	0.05	0.12

Prior result: mercury

- Photochemical conversion from Hg^0 can result in $>200 \text{ pg m}^{-3}$ reactive gaseous mercury (RGM) at 3 km, matching measurements in Florida
(Sillman et al., 2007, submitted)

Hg chemistry incorporated into CMAQ

(Lin and Pehkonen, 1997, 1998; Val Loon et al., 2000;
Sommar et al., 2000, Lindberg, 2002; Khalizof, 2003,
chlorine from JPL 2003, bromine from Sander 1996)

Hg0-to-HgII

(mainly gas-
phase)

- $\text{Hg0} + \text{O}_3 \Rightarrow \text{HgO}$
- $\text{Hg0} + \text{CL}_2 \Rightarrow \text{HgCL}_2$
- $\text{Hg0} + \text{OH} \Rightarrow \text{HgOH}$
- $\text{Hg0} + \text{OH(aq)} \Rightarrow \text{Hg}^{2+}$
etc.

HgII-to-Hg0

(aqueous phase)

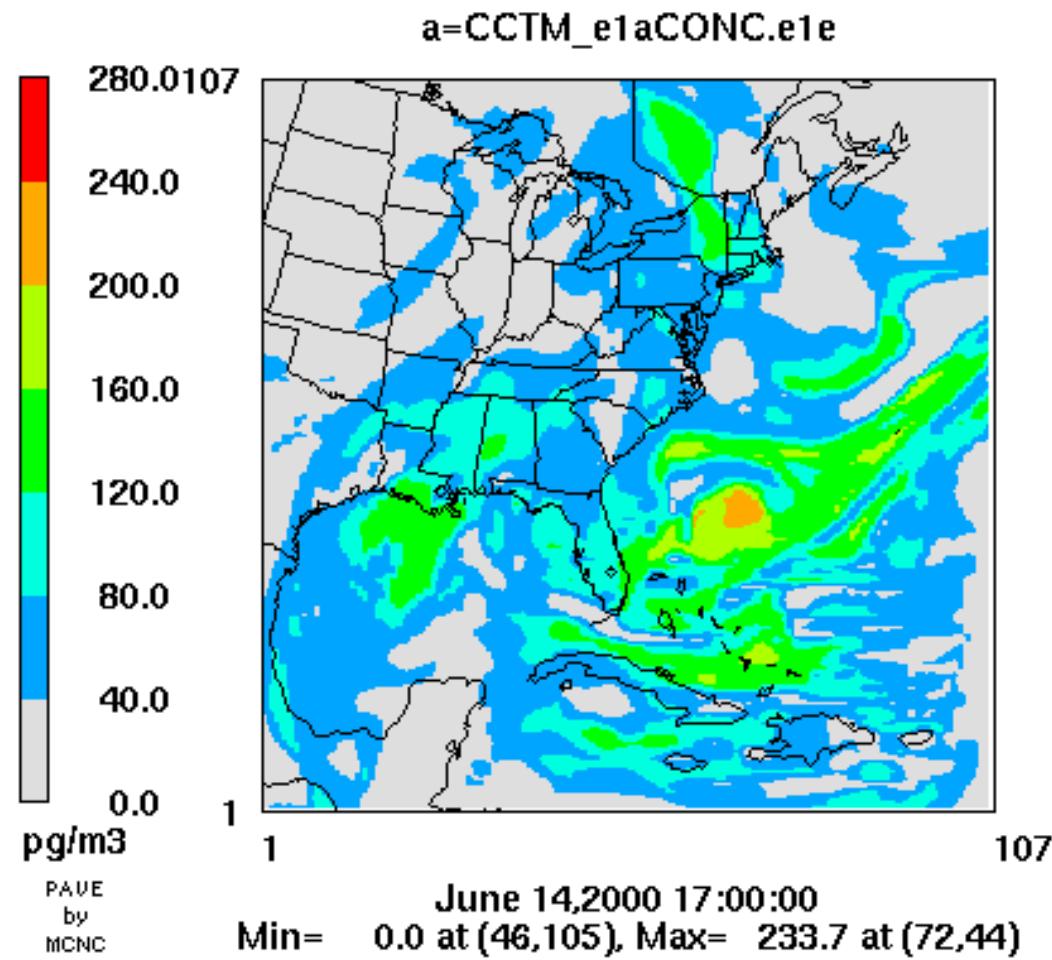
- $\text{Hg}^{2+} + \text{HO}_2(\text{aq}) \Rightarrow \text{Hg0}$
etc.

**(integrated
gas/aqueous/aerosol
chem. is crucial)**

RGM at 3000m

Sillman et al, JGR, submitted

Layer 4 RGMa

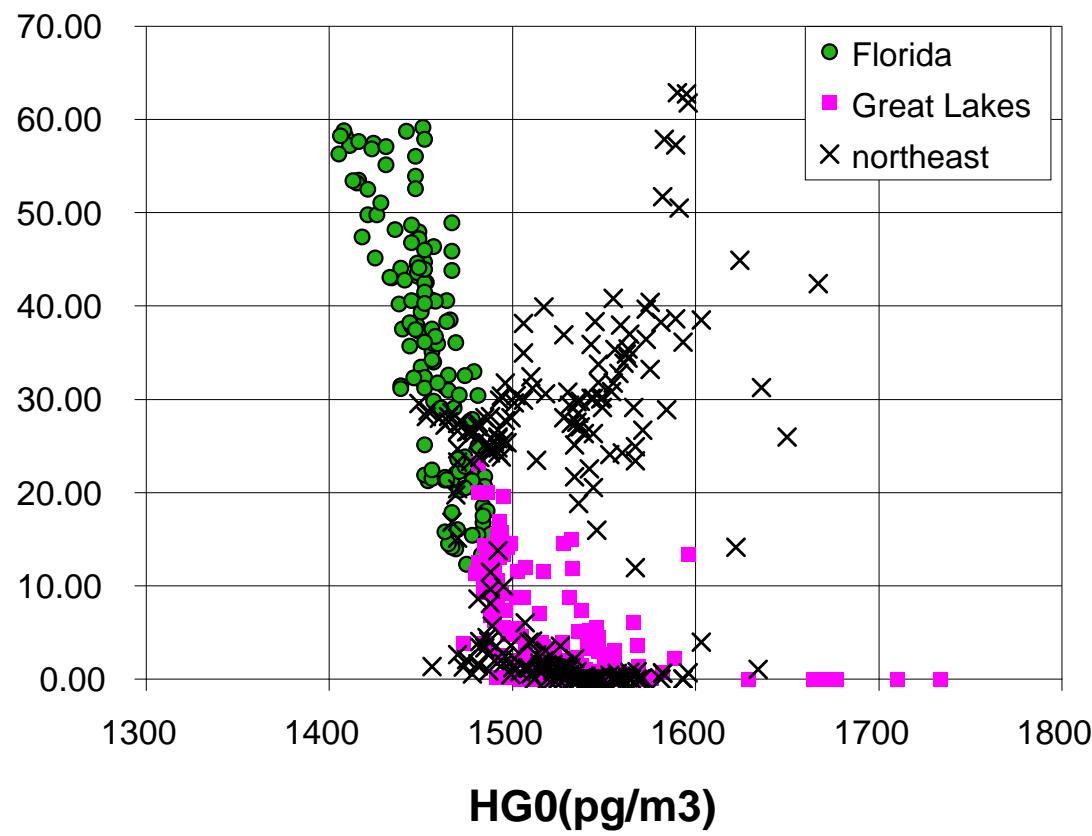


Prior result: mercury

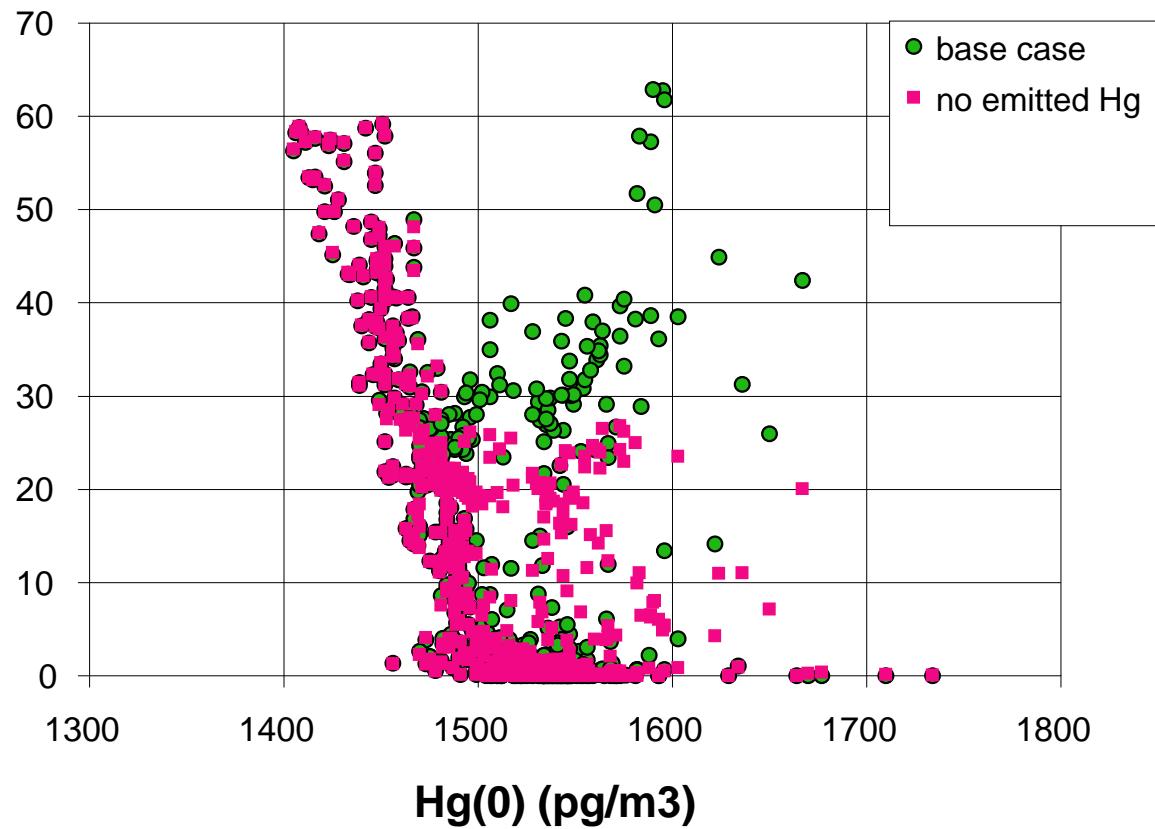
- Correlation between ambient RGM and Hg⁰ can identify global background RGM (negative correlation, model for Florida) versus RGM from emissions (positive correlation, model for northeast)

(WARNING: Results do not preclude impact from local Hg in Florida)

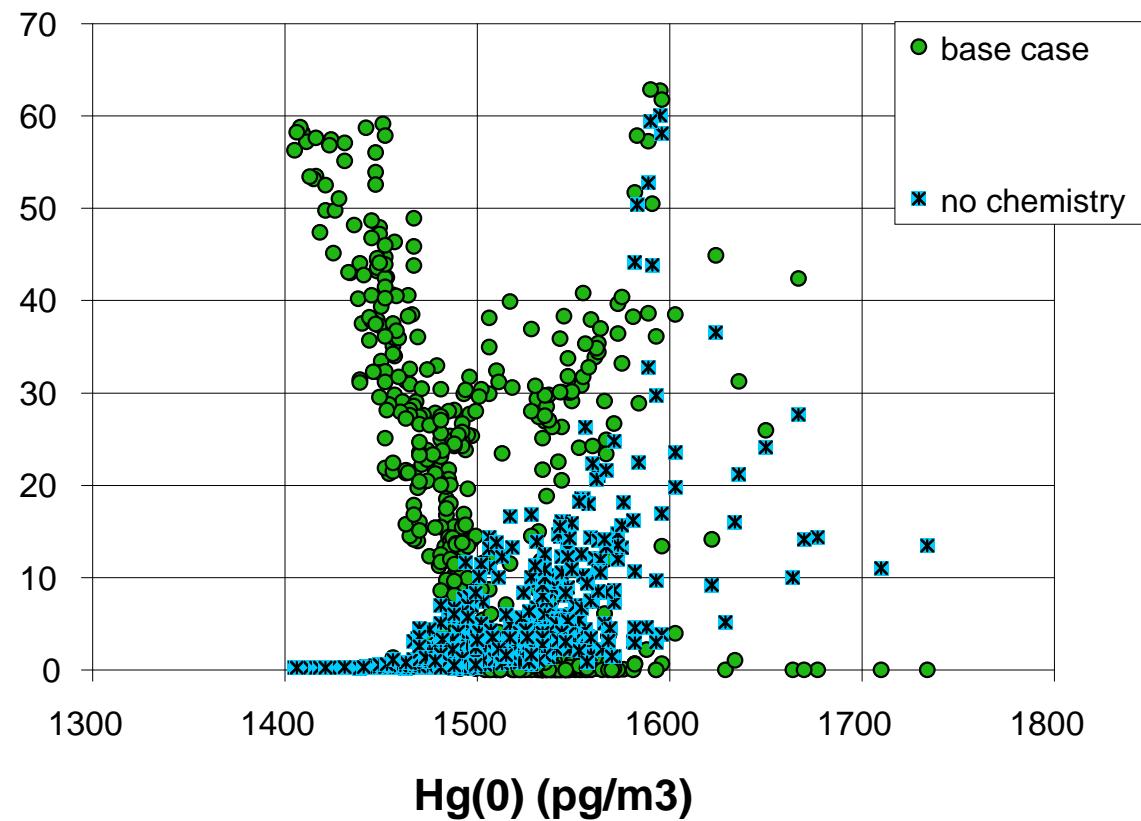
RGM vs Hg0



RGM vs Hg0: zero emissions



RGM vs Hg0: zero background Hg

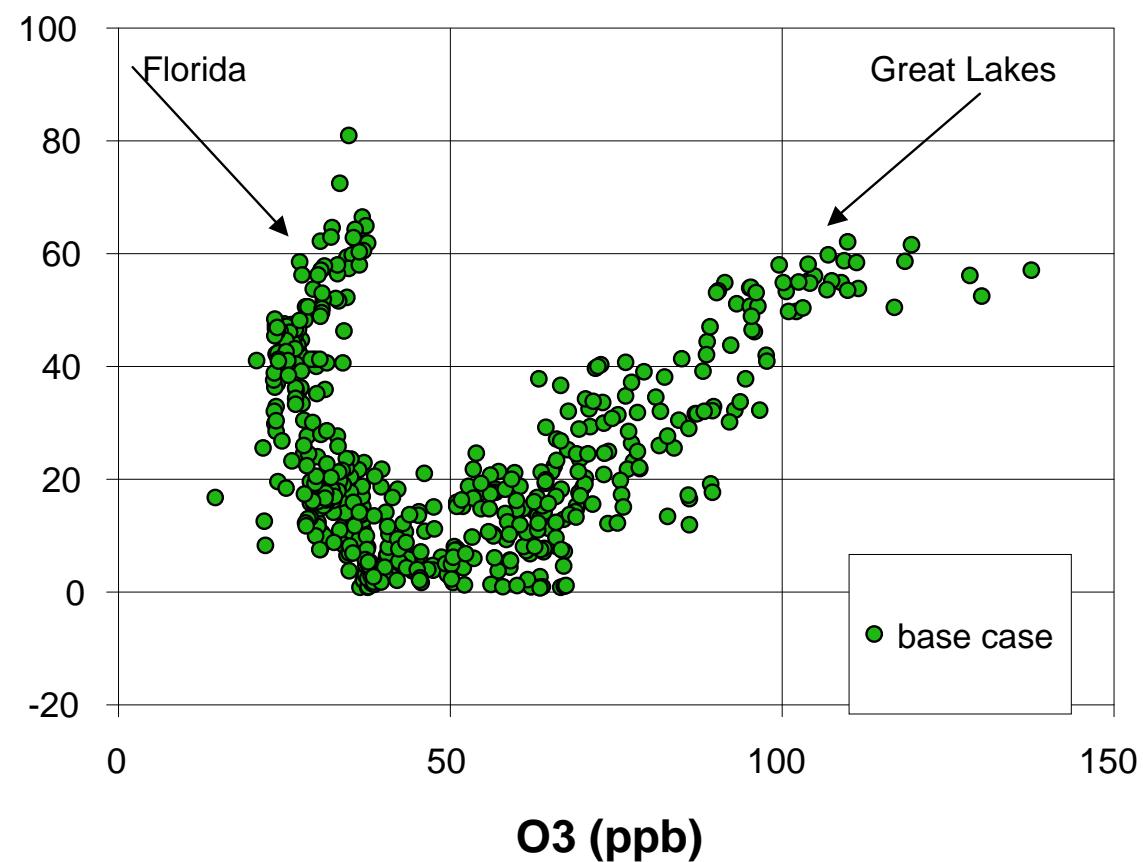


Prior result: O₃ and mercury

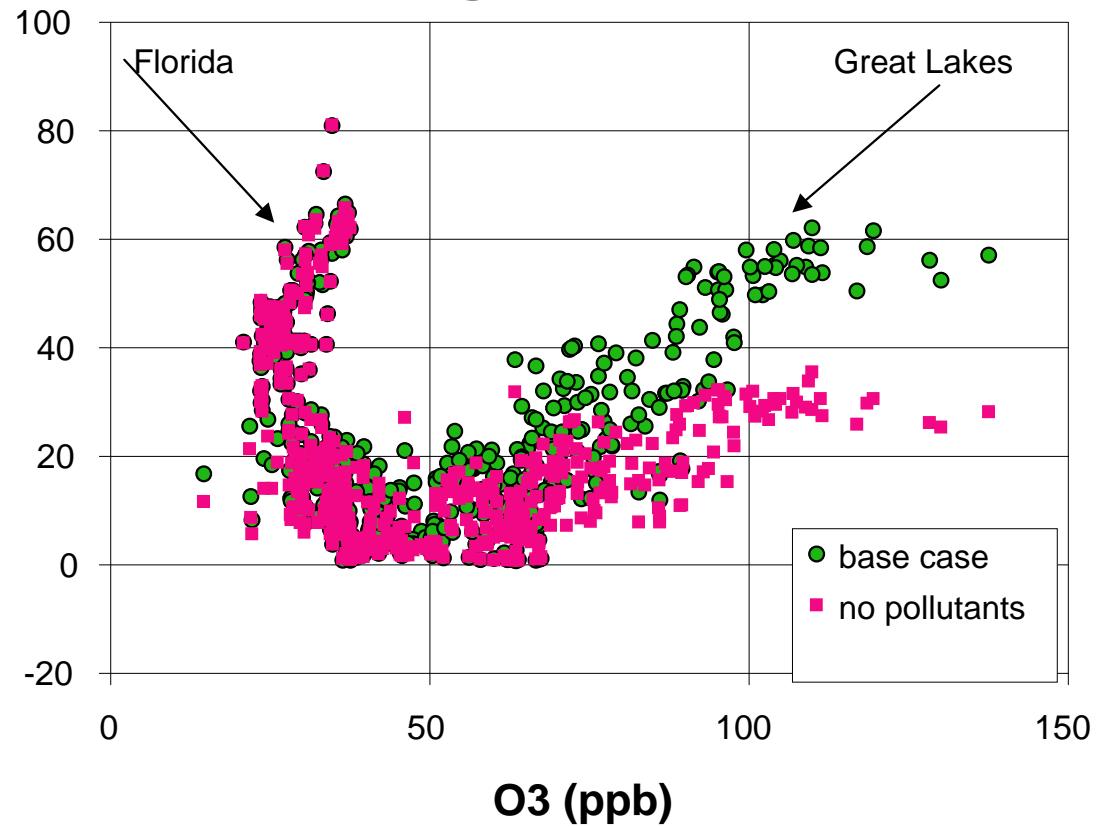
- Models predict a correlation between ambient O₃ and RGM during region-wide pollution events
- The correlation is largely due to simultaneous photochemical production of O₃ and RGM, both resulting from anthropogenic precursors of O₃

Model ambient RGM vs O₃

(June 14, 2000)



Model RGM vs O₃ -with influence of anthropogenic O₃ on RGM removed



Prior result: Hybrid dynamic aerosol model (Feng and Penner, 2007)

- Hybrid dynamic model for nitrate-ammonium aerosols (in place of standard gas-aerosol equilibrium)
- Used in global IMPACT model
- Result: 25% higher fine-mode NO_3^- aerosols, reduced coarse mode, higher gas-phase HNO_3 .
(mainly in remote locations)

Hybrid dynamic model for aerosol NO_3^-

Fine NO_3^- , coarse NO_3^- , HNO_3g
(TgN global)
Equilibrium vs H. dyn. model

Fine mode nitrate aerosol
(pptv)

QuickTime™ and a
TIFF (LZW) decompressor
are needed to see this picture.

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For future investigation:

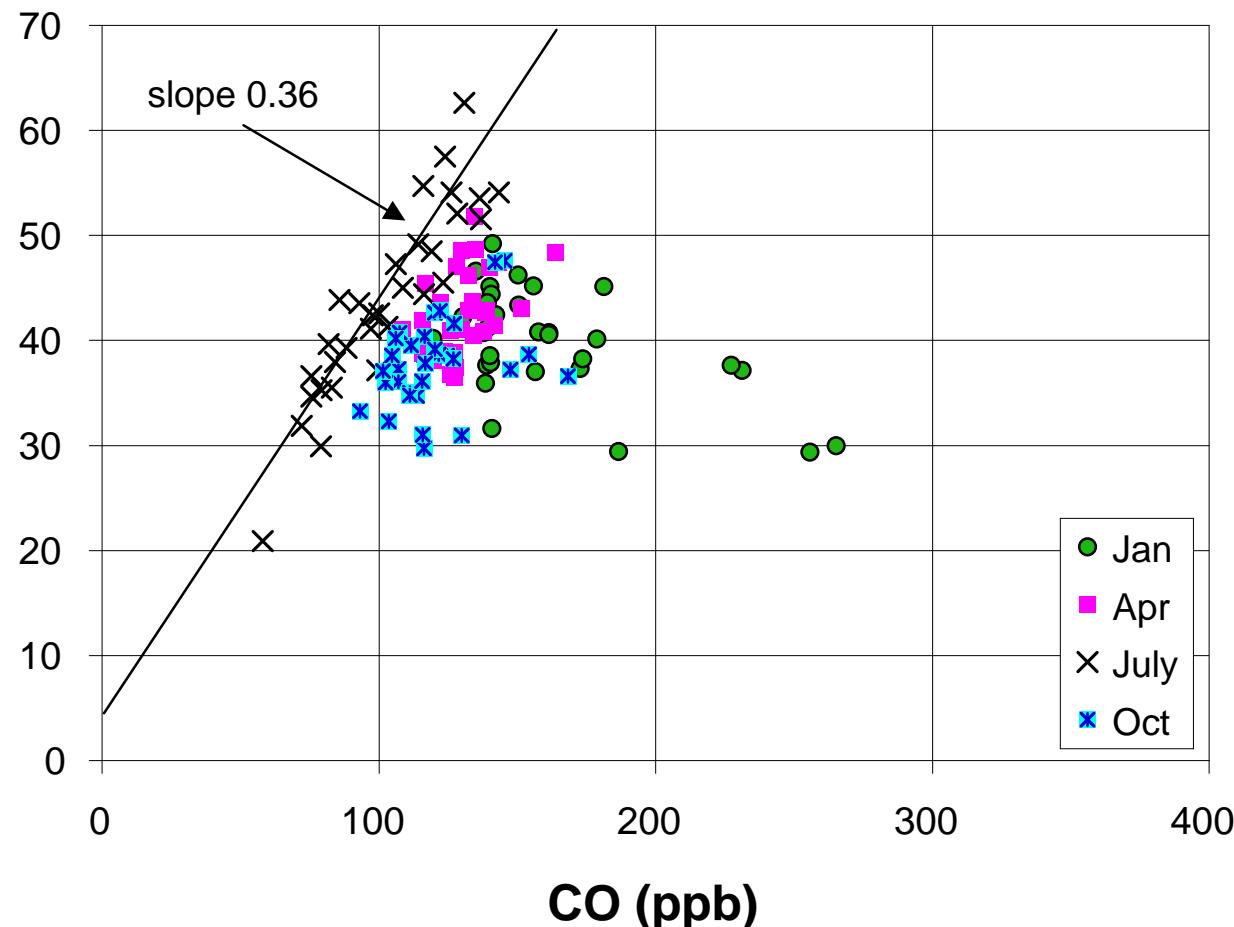
Can species correlations provide a measurement-based signal for the impact of global-scale processes on local O₃?

- (O₃-CO-PAN-HNO₃-H₂O₂)
- (Including episodic transport and the changing global background)

Model O₃ vs CO

Sable Island: 4 seasons

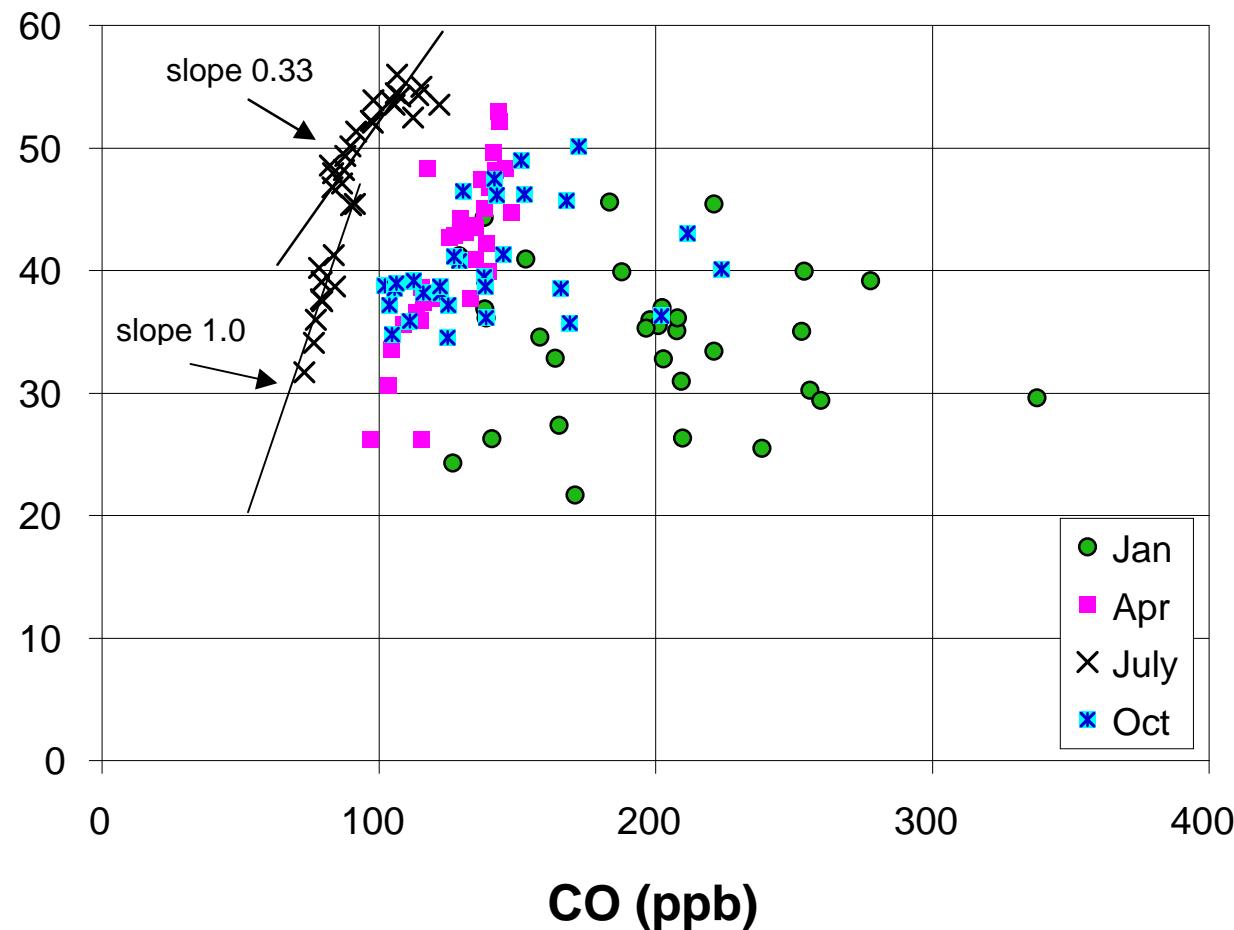
summer slope 0.36: Episodic transport?



Model O₃ vs CO

California/Sierra Nevada

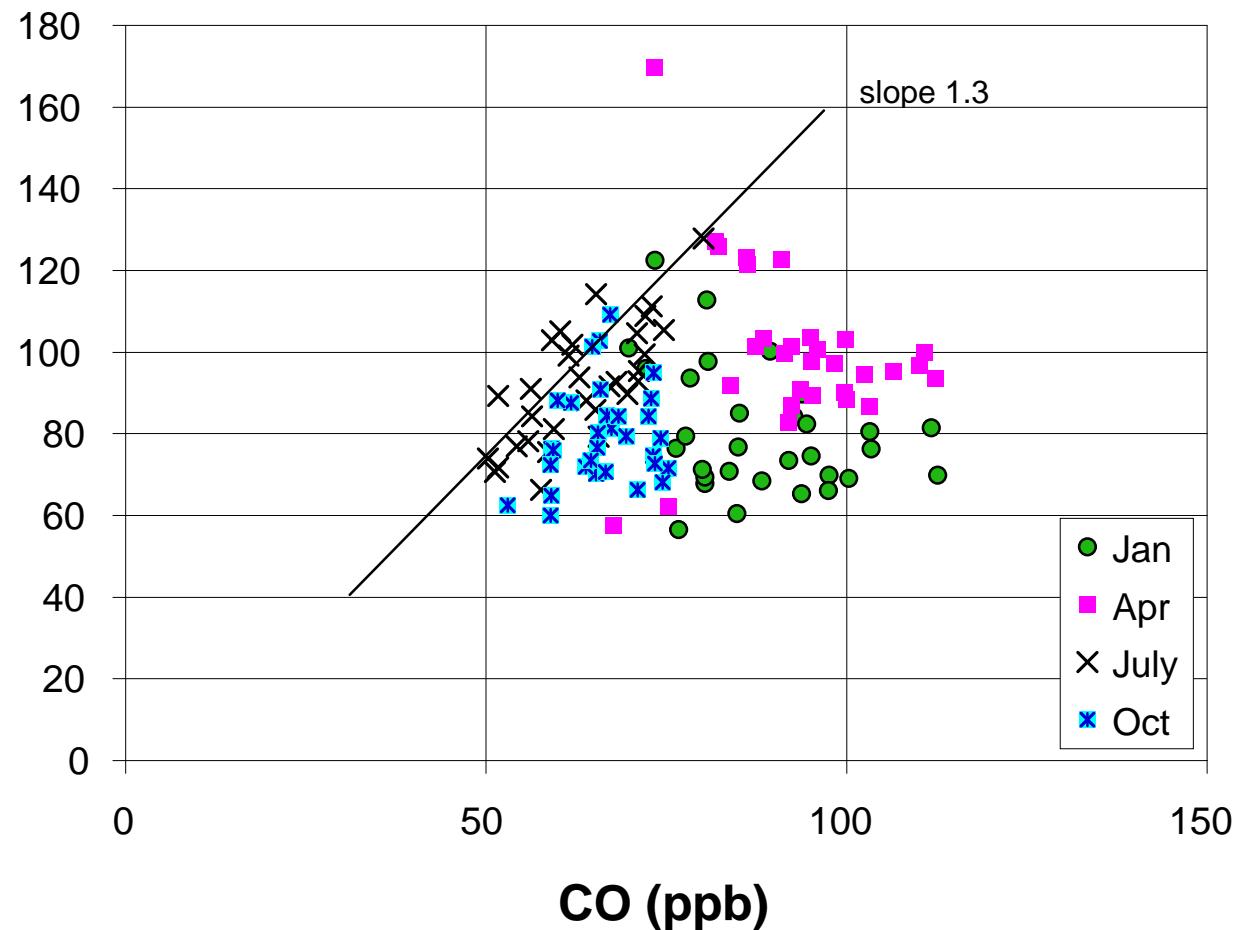
slope >1.0 = global background?



Model O₃ vs CO

California/Sierra Nevada, 50 kPa

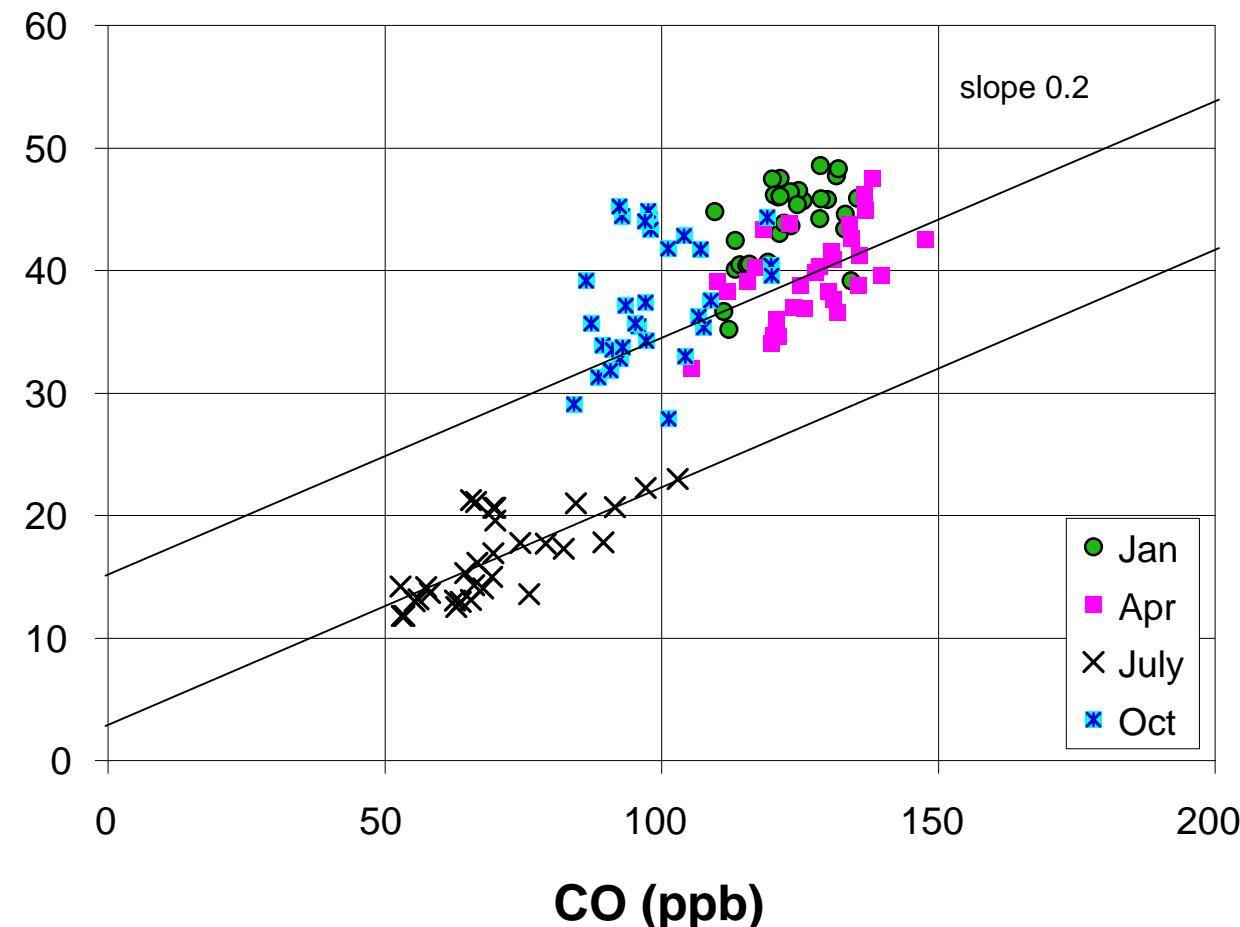
slope >1 is common at 50 kPa



Model O₃ vs CO

Aleutians

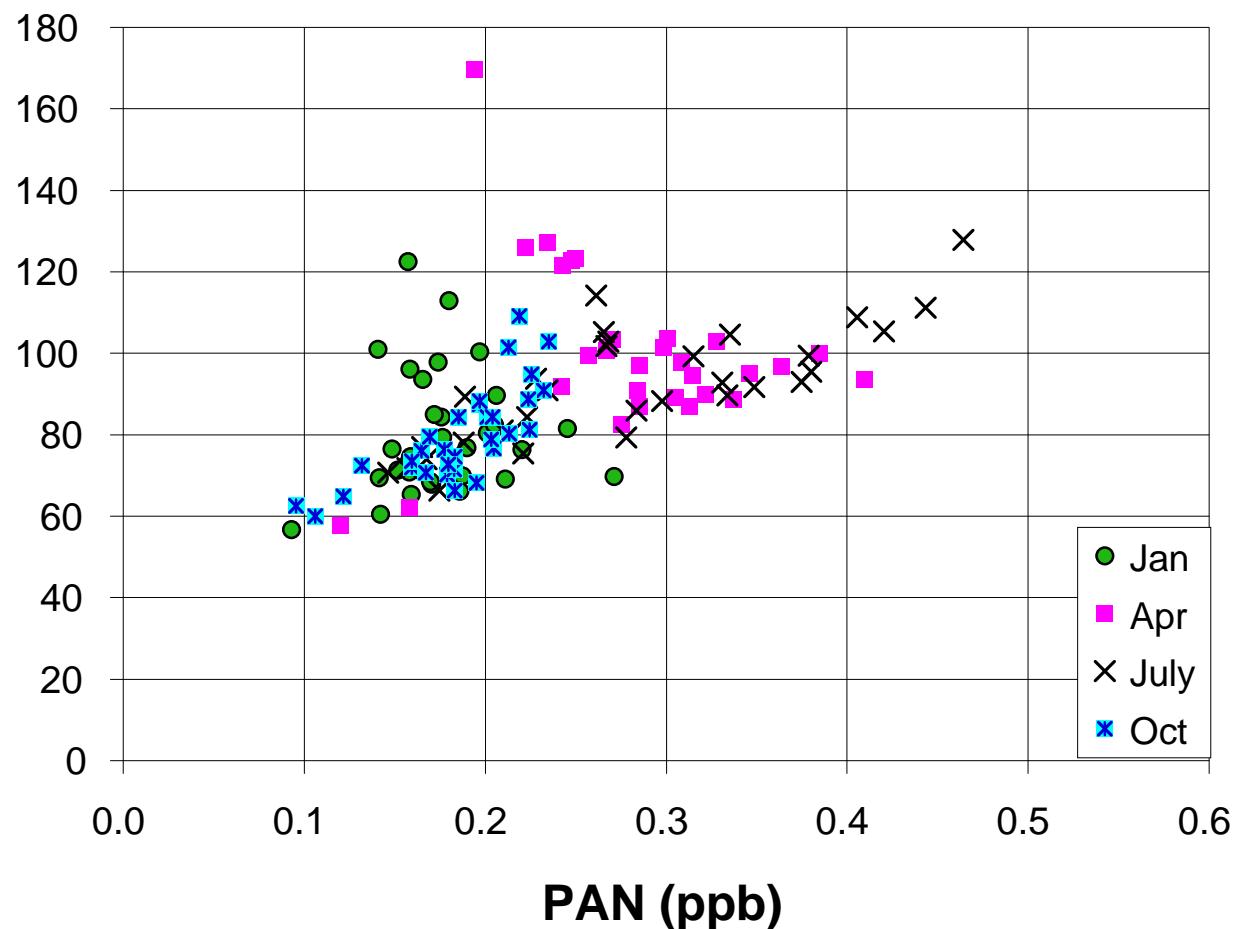
slope 0.2=episodic transport w/O₃ loss?



Model O₃ vs PAN

California/Sierra Nevada, 50 kPa

(+ vs - correlation = different origin?)



Tasks

- CMAQ: link gas, aqueous and aerosols
- Global model: add aqueous/Hg chem.
- Present-day global scenario
- Present day CMAQ - episodes in eastern US.
- Model evaluation vs. measured O₃, CO, NO_y, aerosols and Hg (Keeler)
- Global pattern of Hg wet deposition
- Regional/global simulations for 2050.
- Identify changes in O₃, CO, PAN, HNO₃, H₂O₂ as identifiers of changed global conditions

Expected results and benefits

- Forecast for global impact on regional air quality (from this and other models)
- Methods to identify local versus global impacts on O₃ and mercury from measurements
- Methods to identify global change impact.
- Investigation of new phenomena: relation between O₃ and RGM; effect of aqueous photochemistry
- Improved model capability (CMAQ)

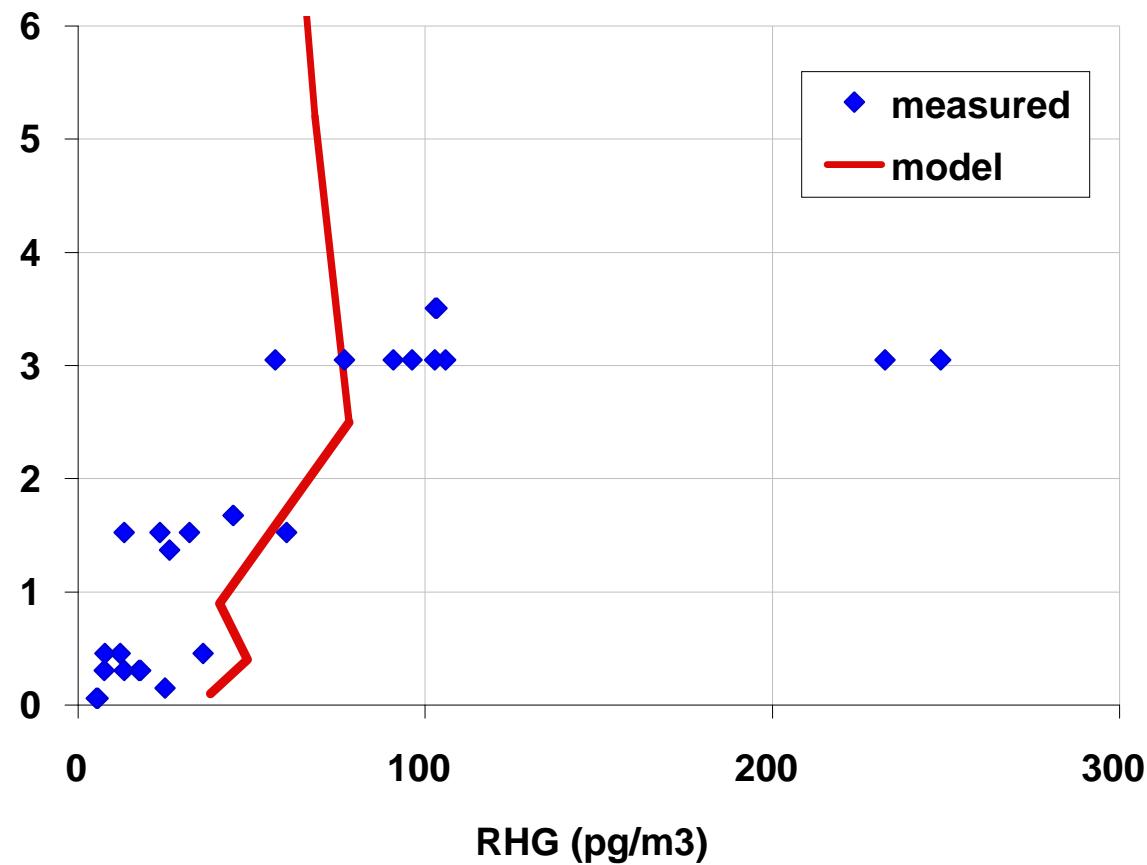


Partial list of references

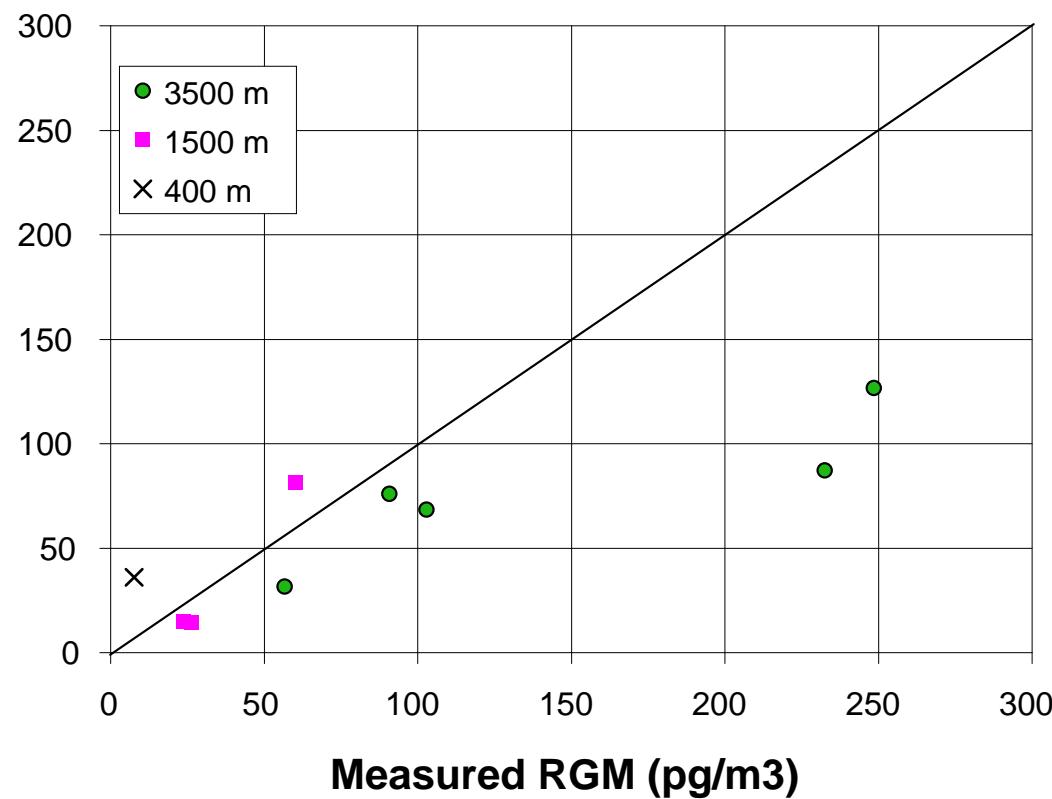
- Barth, M., S. Sillman, R. Hudman, M. Z. Jacobson, C.-H. Kim, A. Monod, J. Liang, Summary of the cloud chemistry modeling intercomparison: Photochemical box model calculation. *J. Geophys. Res.*, 108(D7), 4214, doi:10.1029/2002JD002673, 2003.
- Feng Y., J. E. Penner (2007), Global modeling of nitrate and ammonium: Interaction of aerosols and tropospheric chemistry, *J. Geophys. Res.*, 112, D01304, doi:10.1029/2005JD006404.
- Fiore A. M., L. W. Horowitz, D. W. Purves, H. Levy II, M. J. Evans, Y. Wang, Q. Li, R. M. Yantosca (2005), Evaluating the contribution of changes in isoprene emissions to surface ozone trends over the eastern United States, *J. Geophys. Res.*, 110, D12303, doi:10.1029/2004JD005485.
- Fiore, A. M., D. J. Jacob, I. Bey, R. M. Yantosca, B. D. Field, A. C. Fusco, and J. G. Wilkinson (2002), Background ozone over the United States in summer: Origin, trend, and contribution to pollution episodes, *J. Geophys. Res.*, 107(D15), 4275, doi:10.1029/2001JD000982.
- Ito, A., S. Sillman and J. E. Penner, Effects of additional non-methane volatile organic compounds, organic nitrates and direct emissions of oxygenated organic species on global tropospheric chemistry, *J. Geophys. Res.*, in press, 2006.
- Madronich, S. and S. Flocke, *The role of solar radiation in atmospheric chemistry*, in *Handbook of Environmental Chemistry* (P. Boule, ed.), Springer_Verlag, Heidelberg, pp. 1-26, 1998.
- Spaulding, R.S., G.W. Schade, A.H. Goldstein, and M.J. Charles, Characterization of Secondary Atmospheric Photooxidation Products: Evidence for Biogenic and Anthropogenic Sources, *J. Geophys. Res.*, 108, D8, 4247, doi:10.1029/2002JD002478, 2003.
- von Kuhlmann, R., M. G. Lawrence, U. Pöschl, and P. J. Crutzen (2004), Sensitivities in global scale modeling of isoprene, *Atmos. Chem. Phys.*, 4, 1-17.

Additional slides

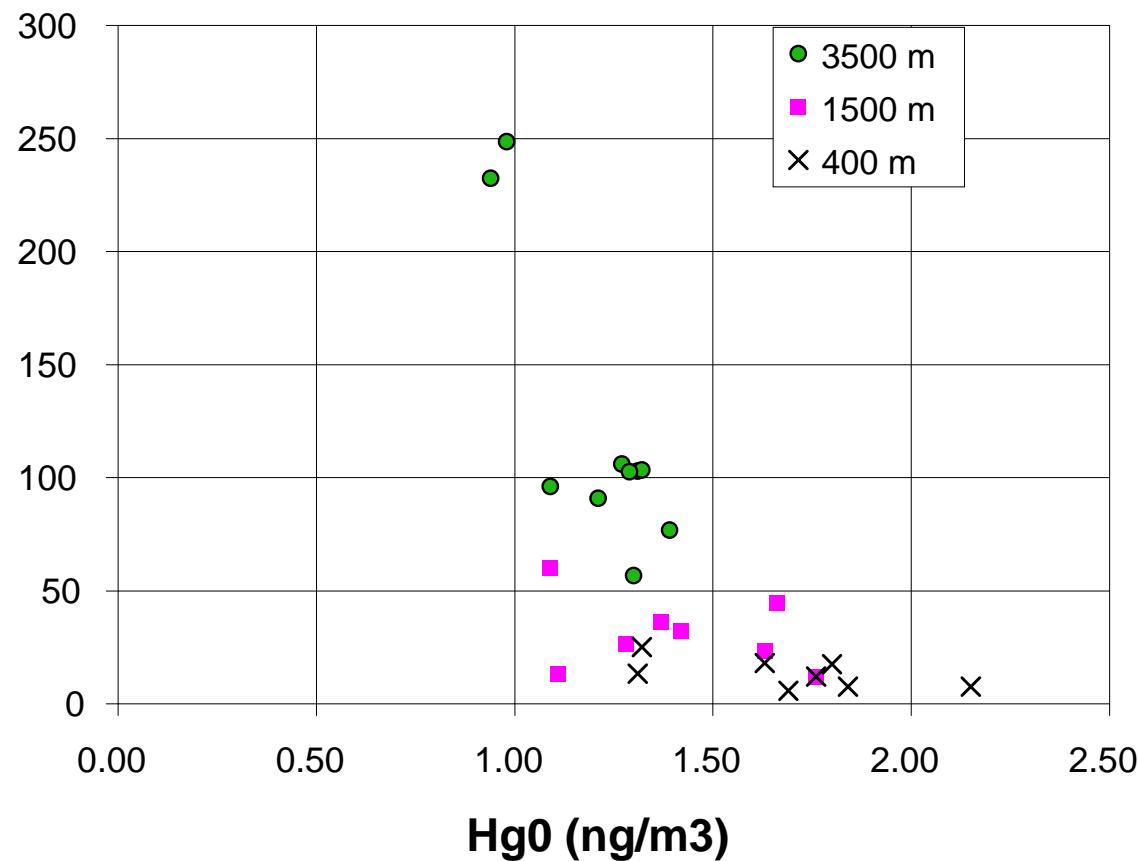
RGM vs altitude: Model and measurements (Landis)



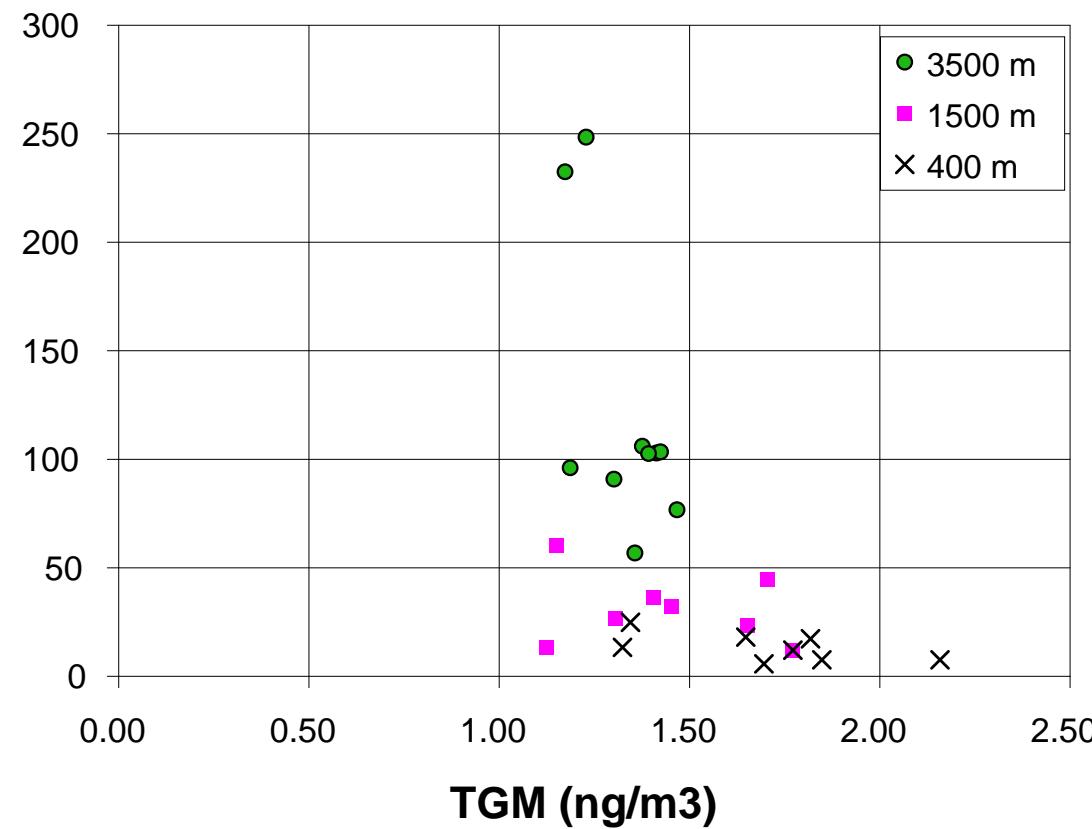
Model vs measured RGM in Florida



RGM v Hg0: Florida June measurements (Landis)

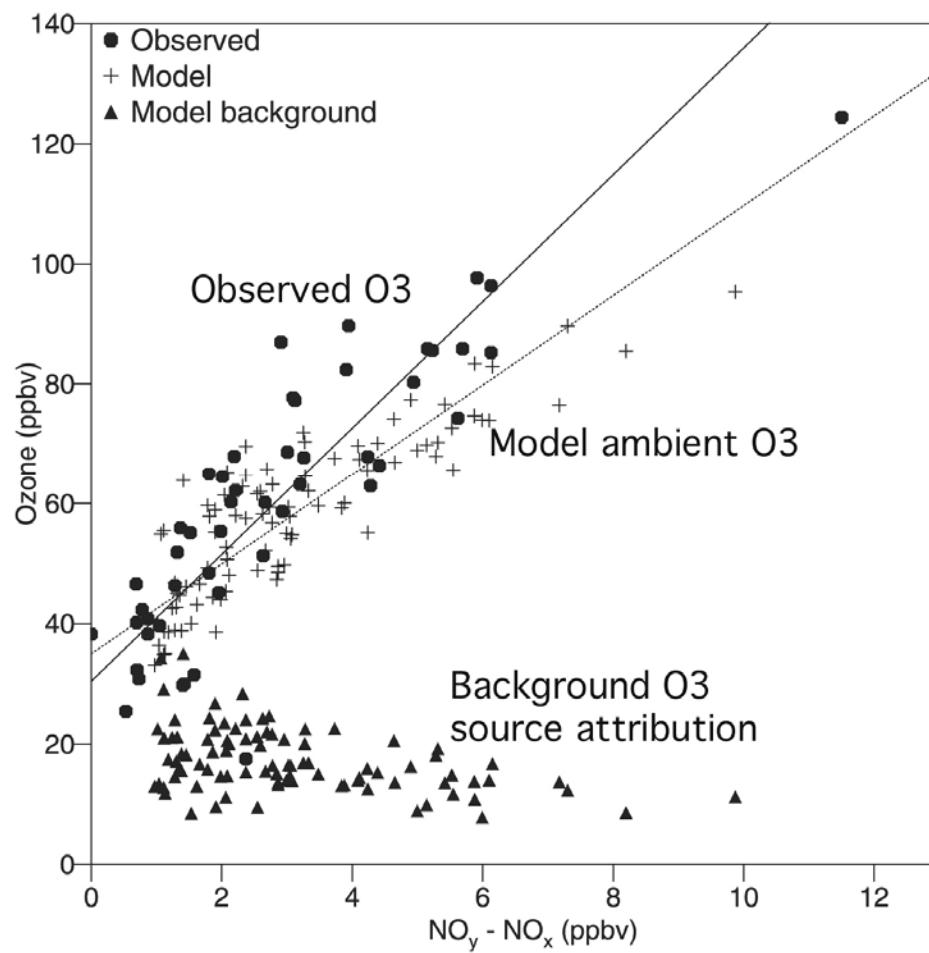


RGM vs Total Hg: June measurements (Landis)



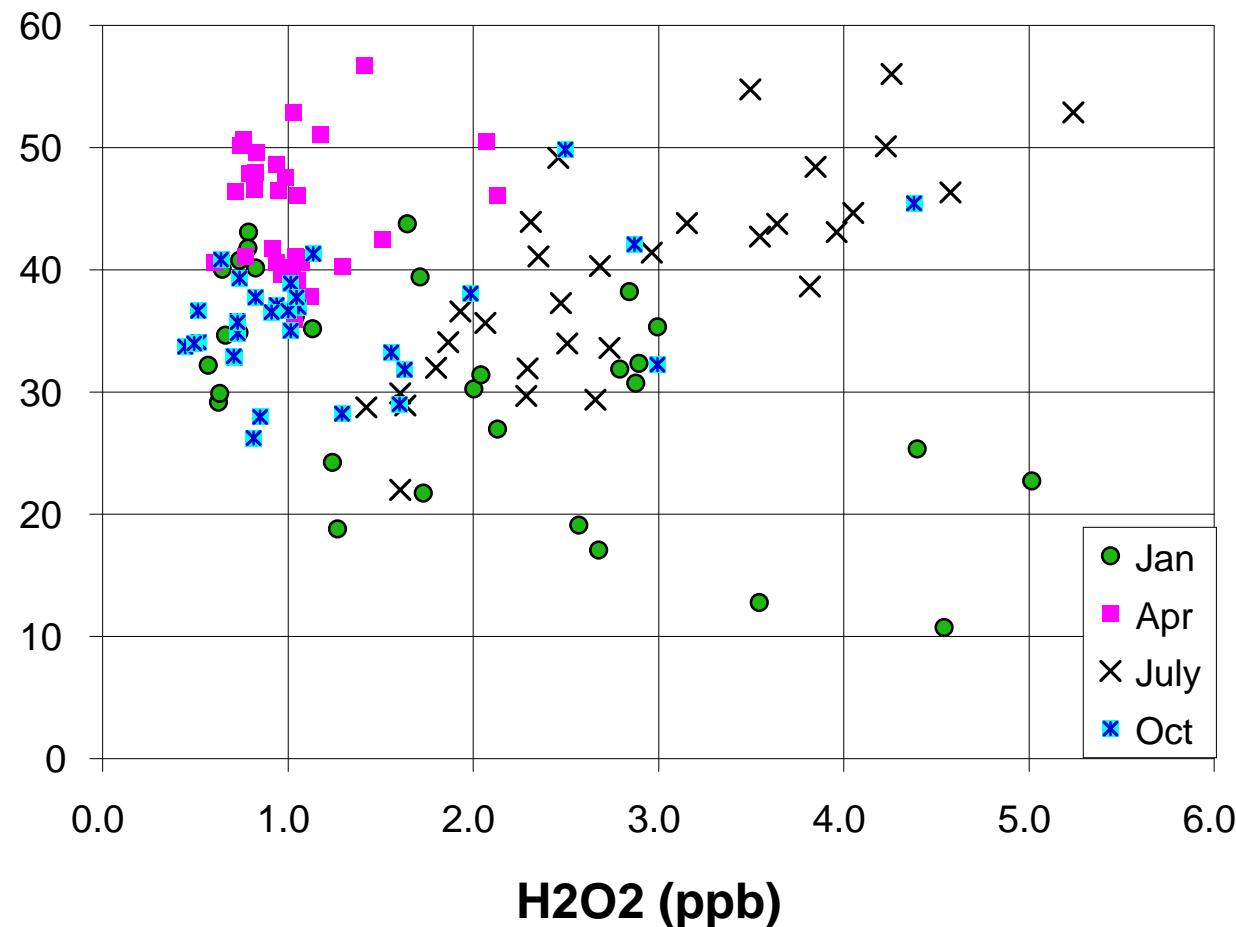
Transported O₃ and source attribution: O₃ vs NO_y (*Fiore et al.*, 2002)

Impact is smaller than inferred global background!



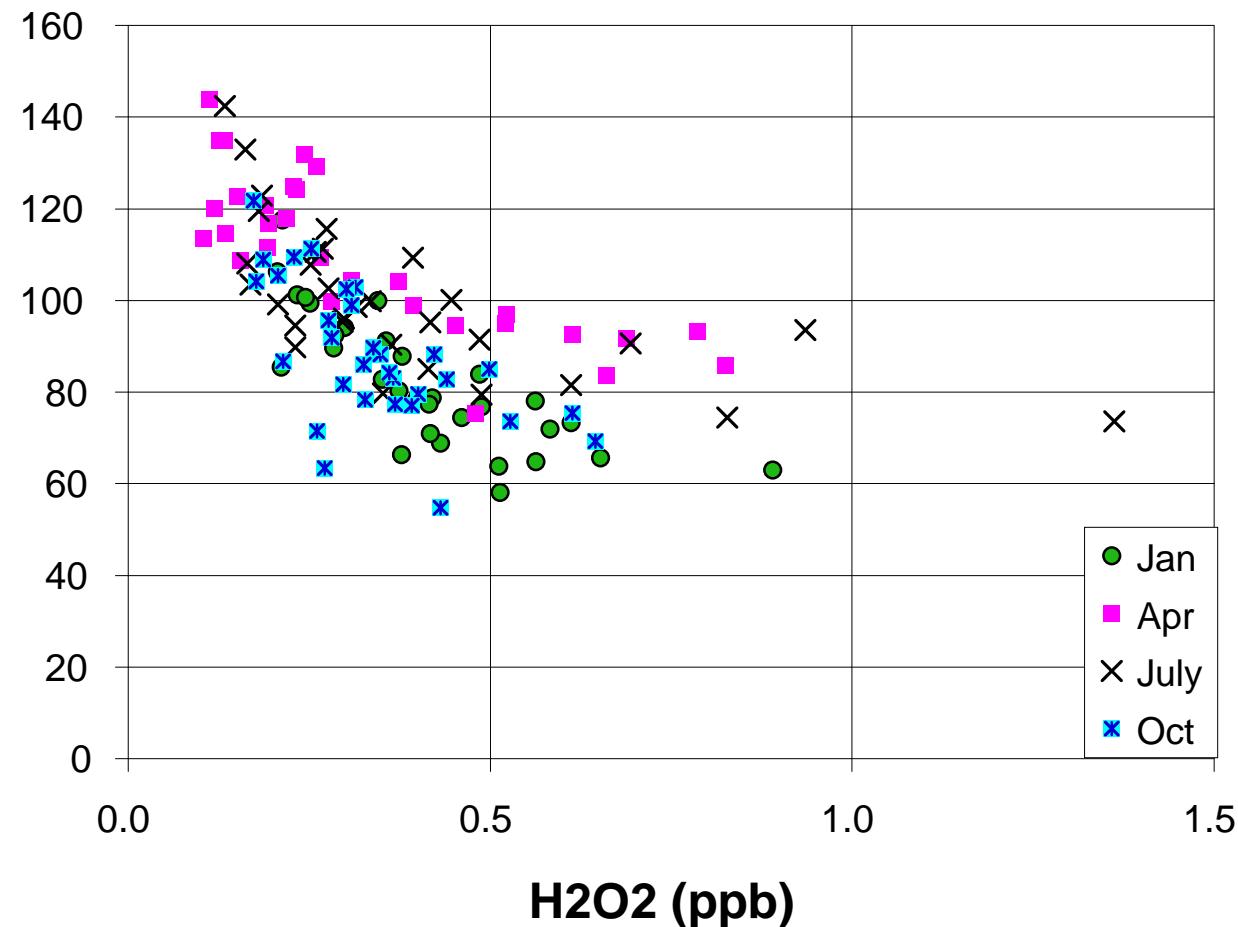
Model O₃ vs H₂O₂

New York/Adirondacks, 99 kPa



Model O₃ vs H₂O₂

New York/Adirondacks, 50 kPa



Conclusions

- Mercury: clear signal for local vs global.
(RGM - Hg⁰ correlation)
- O₃ - CO slopes:
 - 0.36 for regional production of O₃
 - 0.20 for episodic transport
 - 1.0 for global background p'chem
- O₃-HNO₃ at 50 kPa: signal for background p'chem
- Is there an ambient signal for future changes in background O₃? (O₃-NO_y, O₃-H₂O₂?)
- Next: Evaluate with tracers; measurements